FORM PTO-1390 (REV. 11-2000) U.S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE

TRANSMITTAL LETTER TO THE UNITED STATES DESIGNATED/ELECTED OFFICE (DO/EO/US) CONCERNING A FILING UNDER 35 U.S.C. 371

ATTORNEY'S DOCKET NUMBER: GOTEP044

U.S. APPLICATION NO (If known, see 37 CFR 1.5): 09/869365

INTERNATIONAL APPLICATION NO.	INTERNATIONAL FILING DATE	PRIORITY DATE CLAIMED
PCT/SE99/02485	23 December 1999	23 December 1998

TITLE OF INVENTION: GAS DISCHARGE TUBE

APPLICANT(S) FOR DO/EO/US: Schleimann-Jensen, et al.

EL556130861US

Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information:

1. [X] This is a **FIRST** submission of items concerning a filing under 35 U.S.C. 371.
2. [ ] This is a **SECOND** or **SUBSEQUENT** submission of items concerning a filing under 35 U.S.C. 371.
3. [X] This is an express request to begin national examination procedures (35 U.S.C. 371(f)). The submission must include items (5), (6), (9) and (21) indicated below.
4. [ ] The US has been elected by the expiration of 19 months from the priority date (Article 31).
5. [X] A copy of the International Application as filed (35 U.S.C. 371(c)(2))
   a. [ ] is attached hereto (required only if not communicated by the International Bureau).
   b. [X] has been communicated by the International Bureau.
   c. [ ] is not required, as the application was filed in the United States Receiving Office (RO/US).
6. [ ] An English language translation of the International Application as filed (35 U.S.C. 371(c)(2)).
   a. [ ] is attached hereto.
   b. [ ] has been previously submitted under 35 U.S.C. 154(d)(4).
7. [X] Amendments to the claims of the International Aplication under PCT Article 19 (35 U.S.C. 371(c)(3))
   a. [ ] are attached hereto (required only if not communicated by the International Bureau).
   b. [ ] have been communicated by the International Bureau.
   c. [ ] have not been made; however, the time limit for making such amendments has NOT expired.
   d. [X] have not been made and will not be made.
8. [ ] An English language translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371 (c)(3)).
9. [ ] An oath or declaration of the inventor(s) (35 U.S.C. 371(c)(4)).
10. [ ] An English lanugage translation of the annexes of the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371(c)(5)).

Items 11 to 20 below concern document(s) or information included:

11. [ ] An Information Disclosure Statement under 37 CFR 1.97 and 1.98.
12. [ ] An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included.
13. [X] A FIRST preliminary amendment.
14. [ ] A SECOND or SUBSEQUENT preliminary amendment.
15. [ ] A substitute specification.
16. [ ] A change of power of attorney and/or address letter.
17. [ ] A computer-readable form of the sequence listing in accordance with PCT Rule 13ter.2 and 35 U.S.C. 1.821 - 1.825.
18. [X] A second copy of the published international application under 35 U.S.C. 154(d)(4).
19. [ ] A second copy of the English language translation of the international application under 35 U.S.C. 154(d)(4).
20. [X] Other items or information:

Copy of International Preliminary Examination Report

JC18 Rec'd PCT/PTO 25 JUN 2001

U.S. APPLICATION NO. (if known, see 37 CFR 1.5) 09/869365	INTERNATIONAL APPLICATION NO PCT/SE99/02485	ATTORNEY'S DOCKET NUMBER GOTEP044

21. [X] The following fees are submitted:	CALCULATIONS	PTO USE ONLY
BASIC NATIONAL FEE (37 CFR 1.492 (a) (1) - (5)):		
Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO and International Search Report not prepared by the EPO or JPO **$1000.00**		
International preliminary examination fee (37 CFR 1.482) not paid to USPTO but International Search Report prepared by the EPO or JPO **$860.00**		
International preliminary examination fee (37 CFR 1.482) not paid to USPTO but international search fee (37 CFR 1.445(a)(2)) paid to USPTO **$710.00**		
International preliminary examination fee (37 CFR 1.482) paid to USPTO but all claims did not satisfy provisions of PCT Article 33(1)-(4) **$690.00**		
International preliminary examination fee (37 CFR 1.482) paid to USPTO and all claims satisfied provisions of PCT Article 33(1)-(4) **$100.00**		
ENTER APPROPRIATE BASIC FEE AMOUNT =	$ 860.00	
Surcharge of **$130.00** for furnishing the oath or declaration later than [ ] 20 [ ] 30 months from the earliest claimed priority date (37 CFR 1.492(e)).	$	

CLAIMS	NUMBER FILED	NUMBER EXTRA	RATE		
Total claims	20 - 20 =	0	x **$18.00**	$ 0	
Independent claims	2 - 3 =	0	x **$80.00**	$ 0	
MULTIPLE DEPENDENT CLAIM(S) (if applicable)			+ **$270.00**	$	

TOTAL OF ABOVE CALCULATIONS =	$ 860.00	
[ ] Applicant claims small entity status. See 37 CFR 1.27. The fees indicated above are reduced by 1/2. +	$	
SUBTOTAL =	$ 860.00	
Processing fee of **$130.00** for furnishing the English translation later than [ ] 20 [ ] 30 months from the earliest claimed priority date (37 CFR 1.492(f)).	$	
TOTAL NATIONAL FEE =	$ 860.00	
Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31). **$40.00** per property +	$	
TOTAL FEES ENCLOSED =	$ 860.00	
	Amount to be refunded:	$
	charged:	$

a. [X] A check in the amount of $ 860.00 to cover the above fees is enclosed.

b. [ ] Please charge my Deposit Account No. ______ in the amount of $ ______ to cover the above fees. A duplicate copy of this sheet is enclosed.

c. [X] The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. 15-0610. A duplicate copy of this sheet is enclosed.

d. [ ] Fees are to be charged to a credit card. **WARNING:** Information on this form may become public. **Credit card information should not be included on this form.** Provide credit card information and authorization on PTO-2038.

NOTE: Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137 (a) or (b)) must be filed and granted to restore the application to pending status.

SEND ALL CORRESPONDENCE TO

021121

PATENT TRADEMARK OFFICE

Marina T. Larson (signature)

SIGNATURE

Marina T. Larson

NAME

32,038

REGISTRATION NUMBER

GOTEP044

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

U.S. Application No.	To be assigned
Application of:	Schleimann-Jensen, et al
International Application No.	PCT/SE99/02485
International Filing Date:	23 December 1999
Priority Date Claimed:	23 December 1998
For:	Gas Discharge Tube
Attorney Docket No.	GOTEP044

PRELIMINARY AMENDMENT

Asst. Commissioner for Patents

Washington, D.C. 20231

Sir:

Preliminary to calculating the fees for the application filed herewith, please cancel claims 1 - 17 as amended before the IPEA/EP, and add new claims 18 - 37 as follows:

CLAIMS

18. Gas discharge tube comprising at least two electrodes and at least one hollow insulator fastened to at least one of the electrodes, wherein said at least two electrodes have a chemically inert surface, and wherein the chemically inert surface has been arranged onto the electrodes using a physical vapour deposition or a chemical vapour deposition.

19. Gas discharge tube according to claim 18, wherein the chemically inert surface is selected from the group of carbon, gold, and platinum.

20. Gas discharge tube according to claim 19, wherein said carbon is present as polymorph of carbon, such as diamond, diamond-like carbon or graphite.

21. Gas discharge tube according to claim 19, wherein the carbon has been arranged using sputtering.

22. Gas discharge tube according to claim 18, wherein the carbon is arranged in addition of a metal.

23. Gas discharge tube according to claim 18, wherein the metal is chromium or titanium.

24. Gas discharge tube according to claim 18, wherein said carbon is present as polymorph of carbon, such as diamond, diamond-like carbon or graphite.

25. Gas discharge tube according to claim 24, wherein the carbon is present as graphite in addition to a metal.

26. Gas discharge tube according to claim 18, wherein the carbon has been arranged using sputtering.

27. Gas discharge tube according to claim 18, wherein the carbon is present in a layer having a thickness of 1 µm.

28. Method for the manufacture of gas discharge tubes comprising at least two electrodes, and at least one hollow insulator fastened to the electrodes, wherein said at least two electrodes have a chemically inert surface, comprising applying the chemically inert surface onto the electrodes using a physical vapour deposition or a chemical vapour deposition.

29. Method according to claim 28, wherein the chemically inert surface is selected from the group of carbon, gold, and platinum.

30. Method according to claim 29, wherein the carbon is arranged in addition of a metal.

31. Method according to claim 30, wherein the metal is chromium or titanium.

32. Method according to claim 28, wherein said carbon is present as polymorph of carbon, such as diamond, diamond-like carbon or graphite.

33. Method according to claim 32, wherein the carbon is present as graphite in addition to a metal.

34. Method according to claim 28, wherein the carbon has been arranged using sputtering.

35. Method according to claim 28, wherein the deposition of carbon takes place in an atmosphere of methane.

36. Method according to claim 28, wherein the carbon is present in a layer having a thickness of 1µm.

37. Method according to claim 29, wherein said carbon is present as polymorph of carbon, such as diamond, diamond-like carbon or graphite.

REMARKS

These amendments are made to eliminate multiple dependency. No new matter has been added.

Respectfully,

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TITLE

GAS DISCHARGE TUBE

DESCRIPTION

Technical field

The present invention concerns the field of gas discharge tubes including surge arresters, high-intensity discharge tubes, spark gaps and triggered spark gaps, used in various applications, such as surge voltage protectors for communications networks and in particular to a new type of such devices which exhibit higher selectivity, better performance and are more environmentally friendly.

Background of the invention

When electronic equipment is connected to long signal or power lines, antenna etc, it is exposed to transients generated by induction, caused by lightning or electromagnetic pulses (EMP). A surge arrester protects the equipment from damage by absorbing the energy in the transient or by connecting it to ground. Surge arresters are required to be self-recovering, able to handle repetitive transients and can be made fail-safe. An important property is the speed and selectivity of ignition, in other words, the surge arrester must function without delay and still not be so sensitive, that it is triggered by a normal communications signal. These properties should remain unchanged over time and irrespective of the ignition intervals. Further, a surge arrester should be suitable for mass production with high and uniform quality.

Gas-filled discharge tubes are used for protecting electronic equipment but are also frequently used as switching devices in power switching circuits, e.g. in automotive products such as gas-discharge headlights. Other application areas are tele- and data communications, audio/-video equipment, power supplies, industrial, medical devices, security and military applications.

Early surge arresters comprised two solid graphite electrodes, separated by an air-gap or a layer of mica. These are, however, not comparable to the modern surge arresters with respect to size, reliability, performance and production technology.

A modern conventional surge arrester is the gas filled discharge tube, which may have one or several discharge paths or discharge gap and usually comprises two end electrodes plus optionally one additional electrode in the form of a center electrode plus one or two hollow cylindrical insulators, made of an electrically insulating material, such as a ceramic, a suitable polymer, glass or the like. As a rule, the insulator in a two-electrode surge arrester is soldered to the end electrodes at two sides, joining them vacuum tight.

One method of producing a conventional surge arrester is outlined, for example, in US-A- 4,437,845. According to US-A- 4,437,845, the manufacturing process consists of sealing at a suitable temperature the components of the tube at substantially atmospheric pressure in a light gas mixed with another gas which, in view of the intended function of the tube, is desirable and heavier than the first-mentioned gas, and reducing the pressure exteriorally of the tube below atmospheric pressure, while simultaneously lowering the temperature to such extent that the heavy gas can only to an insignificant degree penetrate the tube walls through diffusion and/or effusion, and the enclosed light gas can diffuse and/or be effused through the walls such that, as a result of the pressure difference, it will exit through the walls of the tube, thus causing a reduction in the total gas pressure inside the tube.

Further, an outside coating of the surge arrester components has been disclosed in US-A- 5,103,135, wherein a tin coating is applied to the electrodes, and an annular protective coating is applied to the ceramic insulator having a thickness of at least 1 mm. This protective coating is formed from an acid-resistant and heat-resistant colorant or varnish which is continuous in the axial direction of the surge arrester. The protective coating may form part of the identification of the surge arrester. For example, the identification may be in the form of a reverse imprint in the protective coating. In addition, tin-coated leads can be coupled to the electrodes.

US-A-4,672,259 discloses a power spark gap for protection of electrical equipment against overvoltages and having high current capacity, which spark gap comprises two carbon electrodes each having a hemispherical configuration and an insulating porcelain housing, whereby the carbon electrodes contains vent holes to the inner thereof to provide arc transfer to an inner durable electrode material. The spark gap is intended for high voltage lines, wherein the expected spark length is about 2.5 cm (1 inch), transferring 140 kV or so. This spark gap is not of the type being hermetically sealed and gas filled, but communicates freely with the air. The arc formed starts from the respective underlying electrodes and passes the vent holes. Thus the formation of the spark is, to a great part, based on the underlying material, which is not necessarily inert, but is due to oxidation in the existing environment, which means that the spark voltage can not be determined, and reproduced.

US-A-4,407,849 discloses a spark gap device and in particular a coating on the electrodes of such spark gap, in order to minimize filament formation. The coating is applied onto an underlying electrode, whereby the coating may consist of carbon in the form of graphite. The surge limiter is a gas filled one. The reference does not address the issue of having an inert surface or not on the electrode, or any problems related thereto.

The previously mentioned problems of sensitivity and recovery have been addressed by the use of an electron donor on the electrode surfaces or elsewhere. This electron donor can comprise radioactive elements, such as tritium and/or toxic alkali metals, such as barium. It is obvious, that this solution has specific drawbacks associated *inter alia* with the radioactivity and/or toxicity of the components.

The object of the invention

The object of the present invention is to make available gas discharge tubes for all relevant areas of application, said gas discharge tubes exhibiting higher selectivity, better performance (e.g. higher heat-resistance and longer life-cycle time), and being free of radioactive or otherwise environmentally harmful compounds.

This object is achieved by preventing the build-up of any layers, such as oxide or hydride layers on the electrode surface, in particular on the opposite surfaces of the end electrodes. It is assumed that the formation of oxides on the surface of the metal electrodes influences the onset voltage of a discharge. Regardless of the high vacuum in the discharge chamber, a residue of oxygen and other elements always remains. By preventing layer-formation or oxidation of the electrode surfaces, the discharge tube will repeatedly function at the same voltage or at least within a more narrow interval.

Short description of the drawings

The invention will be described in closer detail below, with reference to the drawings, in which

Fig. 1 shows a cross section of a typical gas discharge tube with two electrodes, and

Fig. 2 shows a cross section of a gas discharge tube with multiple electrodes.

Detailed description of the invention

A generic gas discharge tube comprises at least two electrodes, joined to a hollow insulator body. One frequently encountered type of gas discharge tubes such as illustrated in fig. 1 comprises two end electrodes 1 and 2, each electrode including a flange-like base part and at least one hollow cylindrical insulator 3, soldered to the base part of at least one of the end electrodes. The inventive coating or element, resistant to the build-up of layers, is illustrated as the screened area 4 on both electrodes. Regardless of the type of gas discharge tube, it is important that at least the cathode has the inventive layer or is of the inventive material or construction, which is described below. It is however preferred that all electrodes have this layer or construction, as the polarity of the transient can vary.

The multiple electrode tube illustrated in fig. 2 comprises, in addition to the elements described above, also a centre electrode 5. The inventive coating or element is also here illustrated as a screened area 4, appearing on all electrodes.

It is preferred, that at least part of the opposite surfaces of said end electrodes are covered with a layer or coating of a compound or element, resistant to the build-up of layers, such as

oxide layers. Other unwanted layers, the formation of which the inventive concept aims to prevent, are for example hydrides. In general, the expression "unwanted layers" comprises any layers formed on the electrodes through interaction with surrounding compounds, such as gases contained in the gas discharge tube and which layers influence the performance of the tube.

This compound, which forms the inventive layer and is resistant to the build-up of unwanted layers, can be a highly stable metallic alloy, a metal such as titanium, or a practically inert element, such as gold. The compound can be a carbonaceous compound, preferably carbon with an addition of a metal, such as chromium or titanium.

In this context, carbon is defined as any polymorph of carbon, for example diamond, diamond-like carbon or graphite. The carbon may also contain other elements, such as one or several metals in amounts depending on the application, for example amounts up to about 15 %.

Preferably, the opposite surfaces of said end electrodes are covered with a coating or layer of graphite, said layer comprising an addition of metal, such as chromium or titanium. According to one embodiment of the invention, the inert surface or oxidation resistant coating or layer is applied to the electrodes by chemical plating, sputtering or the like. Preferably, the oxidation resistant layer is applied by conventional sputtering or plasma deposition techniques, well known to a person skilled in the art.

The processes, applicable according to the invention include chemical vapour deposition (CVD), physical vapour deposition (PVD) were a coating is deposited onto a substrate. Sputtering, which is a physical deposition process, is presently held to be the best applicable. In a sputtering process, material is sputtered by bombarding a cathode with high-energetic ions, usually argon ions. When the ions hit the target material, the cathode, atoms will sputter away and deposit onto the substrate. This process generally requires high vacuum or at least low vacuum during the sputtering process. The substrate can be cleaned conveniently by running the process in reverse, by installing the substrate as cathode and bombarding the

same. It is possible to influence the composition of the deposited layer by varying the composition of the gas phase. In an application, where the deposition of a carbonaeceous material is desired, a gaseous hydrocarbon such as methane, can be used. A graphite cathode can also be used as a source of carbon. Using methane together with chromium cathodes, for example, will result in a reactive sputtering process, leading to the deposition of a graphite layer with an addition of chromium. The typical deposition rate is about 1 μm/h or less. Normal sputtering times are in the interval of about 4 to 8 hours. Depending on the desired thickness of the layer, longer or shorter times can be used. By varying the cathode material and the composition of the gas phase, different coatings can be made.

It is also possible, in the case of metallic coatings, to use electroplating procedures or so called electroless plating. These procedures are especially suitable for applying coatings consisting of precious metals, such as gold or platinum.

According to one embodiment of the invention, the surfaces of the electrodes are only partially coated, e.g. on a small area in the direction of the opposite electrode.
As an alternative embodiment of the invention, a part of the electrode is made of the inert material, for example a carbonaceous body, fastened, for example sandwiched or sintered to a metallic base part of the electrode. It is conceived that the electrode can be manufactured as a metallic base, for example a copper or aluminium base, capped with or encasing a graphite body presenting at least one surface in the direction of the at least one opposing electrode.

Surge arresters with electrode surfaces according to the present invention exhibit lower arc voltages and a more narrow distribution of the static ignition voltage than present devices.

Further, the present invention offers a solution, which is easy to implement in existing surge arrester designs, and which is suitable for mass production. Additionally, the solution according to the present invention does not have any negative influence on the environment or require special waste handling procedures, in contrast to presently used surge arresters containing radioactive gas, such as tritium and/or toxic compounds, such barium salts.

Gases used in gas filled surge arresters are i.a., nitrogen, helium, argon, methane, hydrogen, and others, as such or in mixtures.

The invention will be illustrated by a non-limiting production example, which describes the production of a surge arrester according to one embodiment of the invention.

Production example

A surge arrester was produced by subjecting a batch of copper electrodes to the following treatment steps: first, the electrodes were rinsed in a solvent, removing loose contamination and traces of grease or fat. The electrodes were then placed in a mask, exposing the area to be coated. A set of electrodes, cleaned and placed in a mask, were then introduced in a sputtering chamber, which was evacuated. The electrodes were then subjected to cleaning by reverse sputtering, removing impurities from the electrodes. The current was then reversed and methane led into the chamber. By supplying chromium in the form of chromium cathodes, a process of reactive sputtering was performed. The electrodes received a layer of graphite with an addition of chromium atoms locking the graphite layers. Finally, the sputtering process was terminated and the coated electrodes removed from the chamber and subjected to normal quality control.

The coated electrodes exhibited improved qualities, such as higher heat-resistance. Surge arresters manufactured using the coated electrodes exhibited improved qualities, such as lower arc-voltage, more narrow distribution of ignition voltages, and improved speed and selectivity, and longer life-cycle time.

Although the invention has been described with regard to its preferred embodiments, which constitute the best mode presently known to the inventors, it should be understood that various changes and modifications as would be obvious to one having the ordinary skill in this art may be made without departing from the scope of the invention which is set forth in the claims appended hereto.

PCT/SE99/02486

CLAIMS

1. Gas discharge tube comprising at least two electrodes and at least one hollow insulator fastened to at least one of the electrodes, and whereby said at least two electrodes have a chemically inert surface,
characterized in
that the chemically inert surface has been arranged onto the electrodes using a physical vapour deposition or a chemical vapour deposition.

2. Gas discharge tube according to claim 1,
wherein the chemically inert surface is selected from the group of carbon, gold, and platinum.

3. Gas discharge tube according to claim 1,
wherein the carbon is arranged in addition of a metal.

4. Gas discharge tube according to claim 1,
wherein the metal is chromium or titanium.

5. Gas discharge tube according to claim 1-4,
wherein said carbon is present as polymorph of carbon, such as diamond, diamond-like carbon or graphite.

6. Gas discharge tube according to claim 5,
wherein the carbon is present as graphite in addition to a metal.

7. Gas discharge tube according to one or more of claims 1-6,
wherein the carbon has been arranged using sputtering.

8. Gas discharge tube according to one or more of the preceding claims,
wherein the carbon is present in a layer having a thickness of 1 μm.

9. Method for the manufacture of gas discharge tubes comprising at least two electrodes, and at least one hollow insulator fastened to the electrodes, and whereby said at least two electrodes have a chemically inert surface,
characterized in
that the chemically inert surface is applied onto the electrodes using a physical vapour deposition or a chemical vapour deposition.

10. Method according to claim 9,
wherein the chemically inert surface is selected from the group of carbon, gold, and platinum.

11. Method according to claim 10,
wherein the carbon is arranged in addition of a metal.

12. Method according to claim 11,
wherein the metal is chromium or titanium.

13. Method according to claim 9-12,
wherein said carbon is present as polymorph of carbon, such as diamond, diamond-like carbon or graphite.

14. Method according to claim 13,
wherein the carbon is present as graphite in addition to a metal.

15. Method according to one or more of claims 9-14,
wherein the carbon has been arranged using sputtering.

16. Method according to one or more of claims 9-15,
wherein the deposition of carbon takes place in an atmosphere of methane.

17. Method according to claims 9-16,
wherein the carbon is present in a layer having a thickness of 1 µm.

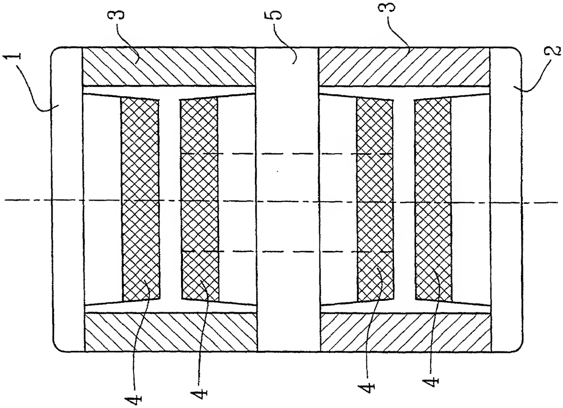

FIG.2

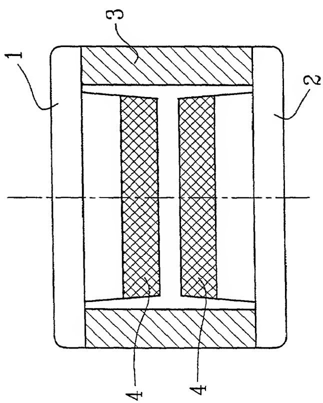

FIG.1

DECLARATION FOR UTILITY OR DESIGN PATENT APPLICATION

As a below named inventor, I hereby declare that:
My residence, post office address and citizenship are as stated below next to my name.

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter, which is claimed and for which a patent is sought on the invention entitled: **"Gas discharge tube"**

the specification of which
[ ] is attached hereto.
[x] was filed on June 25, 2001 as United States Application Number or PCT International Application Number 09/869,365 and was amended on _____ (if applicable).

I hereby state that I have reviewed and understand the contents of the above-identified specification, including the claims, as amended by any amendment specifically referred to above.

I acknowledge the duty to disclose information, which is material to patentability as defined in Title 37, Code of Federal Regulations, § 1.56.

I hereby claim foreign priority benefits under Title 35, United States Code, § 119 (a)-(d) of any foreign application(s) for patent or inventor's certificate or § 365(a) of any PCT international application which designated at least one country other than the United States of America, listed below and have also identified below any foreign application for patent or inventor's certificate, or any PCT international application having a filing date before that of the application on which priority is claimed:

Prior foreign application(s)			
Country	Application Number	Date of Filing (day/month/year)	Priority Claimed
Sweden	**9804538-8**	**23.12.1998**	YES [X] NO [ ]
			YES [ ] NO [ ]

I hereby claim the benefit under Title 35, United States Code, § 119(e) of any United States provisional application(s) listed below:
Application Number: Filing Date:
Application Number: Filing Date:

I hereby claim the benefit under Title 35, United States Code, § 120 of any United States application(s) or § 365(c) of any PCT international application designating the United States of America, listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States or PCT international application in the manner provided by the first paragraph of Title 35, United States Code, § 112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, § 1.56 which became available between the filing date of the prior application and the national or PCT international filing date of this application:

U.S. Parent Application Serial Number: Parent Filing Date: Parent Patent No:

U.S. Parent Application Serial Number: Parent Filing Date: Parent Patent No:

PCT Parent Number: **PCT/SE99/02485** Parent Filing Date: **23.12.1999**

POWER OF ATTORNEY: I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and to transact all business in the Patent and Trademark Office connected therewith:

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I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further, that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

1 - 00

Full name of sole first inventor (given name, family name): **Mats Boman**

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2 - 00

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Inventor's signature: Date: 2001 09/9

Residence: **Trädgårdsvägen 29** Citizenship: **Swedish**

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3 - 00

Full name of third joint inventor (given, name, family name): **Kjell Öhman**

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Residence: **Rosenlundsvägen 9** Citizenship: **Swedish**

Post Office address: **197 34 BRO, Sweden** SE

4 - 00

Full name of fourth joint inventor (given, name, family name): **Johan Schleimann-Jensen**

Inventor's signature: Date: September 20, 2001

Residence: **Ekbacksvägen 33A** Citizenship: **Swedish**

Post Office address: **182 34 DANDERYD, Sweden** SE

☐ Additional inventors are being named on separately numbered sheets attached hereto.